

The Role of Various Coagulant in Water Treatment

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Abstract: Coagulation is the natural and chemical water treatment process is used to remove small suspended solid particles from all types of water. At present time the disposal of the sludge is becoming a major problem. One of the major environmental challenges facing municipalities around the world is the disposal of waste. Therefore, it is necessary to utilize the waste effectively with technical department. This study aims to treat the water by using natural and chemical coagulants. Natural coagulants are watermelon seeds, mooring seeds, and other seeds. Chemical coagulants are alum, sodium aluminate and ferric sulphate. Untreated sample and treated water sample will go through physical, chemical, and biological examination like PH, color, turbidity, TDS, TSS, MPN, etc. This project gives an idea about use of suitable coagulants i.e., natural, and chemical.

Keywords: coagulants, sludge, suspended solids, types of examination on sample.

1. Introduction

A. Types of Coagulants

There are two types of coagulants that are most used in water and wastewater treatment:

- 1) Chemical coagulant
 - Alum coagulant
 - Ferric sulphate
- 2) Natural coagulants
 - Moringa seed
 - Okra seed
- 1) Chemical coagulants

Chemical Coagulation is the process used to eliminate solids from water, by manipulating electrostatic charges of atoms suspended in water.

• Alum coagulant:

Alum is the maximum used as alum coagulant. It is obtainable in several solid forms such as block, kibbled or ground and is also available as a solution. In waterworks practice aluminum sulphate is frequently but imperfectly referred to as 'alum'. [3]

• *Ferric sulphate:*

Ferric Sulphates are effective primary coagulants that are used by the water industry for consumption water production, wastewater treatment requests such as phosphorus removal,

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struvite control and sludge digestion. [4] Ferric Sulphate as Coagulant and the Chemical Reactions like alum, ferric sulphate also needs alkalinity in the water in instruction to form the flock particles ferric hydroxide [Fe (OH)₃]. When natural alkalinity is not enough, alkaline chemicals (such as soluble salts containing, CO^{-2} and OH^{-} ion) should be additional.

2) Natural coagulants

Natural coagulant is a naturally occurred; plants-based coagulant that can be used in coagulation-flocculation procedure of wastewater treatment for plummeting turbidity.

• Moringa seeds:

Numerous studies have been conducted and presented that moringa seeds are real as bio coagulant to improve physiochemical properties of dirty water. The seed extract can separate unwelcome particulates from water sediment impurities. [3] They also have potential as anti-microbial treatment – the unprocessed seed powder may sediment over 90% of the microorganisms from raw water.

• Okra seeds:

Okra is previously an important vegetable crop grown in tropical and subtropics parts of the biosphere. The okra seeds are used for the conduct of water sample. For the water treatment, the oil contained in the okra seeds was first extracted, before the okra seeds used. In the range of studied, it is experiential that whatever the volume of gumbo mucilage, the turbidity reductions when the pH rises. [4]



Fig. 1. Moringa seed powder



Fig. 2. Okra seeds

Types of tests carried out:

- 1. Ph
- 2. Alkalinity
- 3. Hardness
- 4. TDS
- 5. TSS



Fig. 3. Alkalinity test

2. Objectives

- To determine the alkalinity, pH, and hardness test on the surface water.
- To determine the TDS, TSS and MPN test on the surface water.
- Comparison between natural and chemical coagulants in terms of effectiveness and cost.

3. Result	S
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Table 1
Comparison between natural and chemical in terms of effectiveness
1 Alum

	1. Alum				
Sr. no.	Amount of coagulant added	Type of test	Result		
1.		pН	7.8		
2.		Alkalinity	92.3		
3.	1 ~~~	Hardness	29.8		
4.	1 gm	TDS	363		
5.		TSS	3.92		
6.		MPN	35		
	2. Moringa				
1.	1	pН	8.9		
2.	l gm	Alkalinity	203		

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$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	16.	5 gm	TDS	403
3. Okra 1. pH 8.9 2. 1 gm Alkalinity 298 3. 1 gm Alkalinity 298 4. 1 gm Alkalinity 298 4. 1 gm Alkalinity 298 6. PH 8.7 36.9 6. PH 8.7 8. PH 8.7 9. 3 gm PH 8.7 10. 3 gm TDS 430 11. PH 8.7 Alkalinity 260 11. PH 8.7 Alkalinity 260 13. PH 7.5 Alkalinity 97 15. 5 gm TDS 560 TSS 20.89 16. 17. State PH 8.6 Alkalinity 127 3. 1 gm 1 gm TDS 560 TSS 16 6. TSS 16 MPN 53 <td< td=""><td>17.</td><td></td><td>TSS</td><td>15.86</td></td<>	17.		TSS	15.86
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	18.		MPN	76
$ \begin{array}{c ccccc} 2. \\ 3. \\ 4. \\ 1 \ {\rm gm} \end{array} \begin{array}{c} Alkalinity & 298 \\ Hardness & 254 \\ TDS & 430 \\ TSS & 36.9 \\ \hline \\ 6. \\ \end{array} \\ \begin{array}{c} TSS & 36.9 \\ PH & 8.7 \\ \hline \\ 8. \\ 9. \\ 3 \ {\rm gm} \end{array} \begin{array}{c} pH & 8.7 \\ \hline \\ Alkalinity & 260 \\ Hardness & 263 \\ \hline \\ 10. \\ 11. \\ 12. \\ \hline \\ 12. \\ 12. \\ \hline \\ 13. \\ 14. \\ 12. \\ \hline \\ 14. \\ 15. \\ 14. \\ 16. \\ \hline \\ 15. \\ 5 \ {\rm gm} \end{array} \begin{array}{c} pH & 7.5 \\ \hline \\ Alkalinity & 97 \\ \hline \\ Hardness & 25.98 \\ \hline \\ 16. \\ \hline \\ 17. \\ \hline \\ 16. \\ \hline \\ 17. \\ \hline \\ 18. \\ \hline \\ 18. \\ \hline \\ \hline \\ 18. \\ \hline \\ 18. \\ \hline \\ \hline \\ 10. \\ \hline \\ 17. \\ \hline \\ 18. \\ \hline \\ 1. \\ 2. \\ 3. \\ 1 \ {\rm gm} \end{array} \begin{array}{c} pH & 8.6 \\ \hline \\ Alkalinity & 127 \\ Hardness & 60 \\ \hline \\ TDS & 560 \\ \hline \\ TDS & 560 \\ \hline \\ TDS & 560 \\ \hline \\ 17. \\ \hline \\ 1. \\ 2. \\ 3. \\ 1 \ {\rm gm} \end{array} \begin{array}{c} pH & 8.6 \\ \hline \\ Alkalinity & 127 \\ Hardness & 60 \\ \hline \\ TDS & 560 \\ \hline \\ TSS & 16 \\ \hline \\ 6. \\ \hline \\ 7. \\ 9. \\ 3. \\ 1 \ {\rm gm} \end{array} \begin{array}{c} pH & 8.6 \\ \hline \\ Alkalinity & 127 \\ Hardness & 60 \\ \hline \\ TDS & 560 \\ \hline \\ TSS & 16 \\ \hline \\ 6. \\ \hline \\ 10. \\ \hline \\ 10. \\ \hline \\ 11. \\ 1. \\ \hline \\ 10. \\ \hline \\ 11. \\ 1. \\ \hline \\ 10. \\ \hline \\ 11. \\ 11. \\ \hline \\ 10. \\ \hline \\ 11. \\ \hline \\ 10. \\ 0.1 \ {\rm gm} \end{array} \begin{array}{c} Alkalinity & 200 \\ \hline \\ Alkalinity & 200 \\ \hline \\ Alkalinity & 93.5 \\ \hline \\ \hline \\ Alkalinity & 93.5 \\ \hline \\ $		3. Okra		
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8. Alkalinity 200 9. 3gm Hardness 90 10. TDS 600 11. TSS 20 12. MPN 46 13. pH 7.2 14. Alkalinity 93.5 15. 0.1gm TDS 540 17. TSS 3.53				
9. 3gm Hardness 90 10. TDS 600 TSS 20 11. TSS 20 MPN 46 13. pH 7.2 Alkalinity 93.5 15. 0.1gm TDS 540 17. TSS 3.53				
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16. 0.1gm TDS 540 17. TSS 3.53	14.			
10. 1DS 340 17. TSS 3.53	15.	0.1~~~	Hardness	26.1
	16.	0.1gm	TDS	540
18. MPN 39	17.		TSS	3.53
	18.		MPN	39

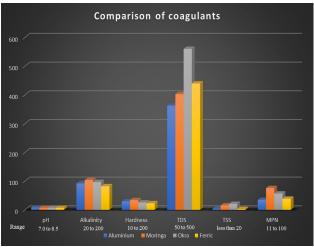
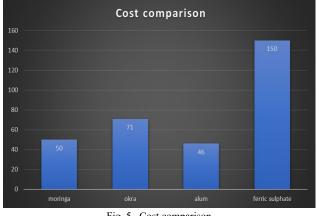
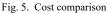


Fig. 4. Comparison of coagulants

Table 2 Comparison between natural and chemical in terms of cost

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Coagulant	Gram	Cost		
Natural coagulants				
Moringa	100	50 Rs.		
Okra	100	71 Rs.		
Chemical coagulants				
Alum	100	46 Rs.		
Ferric sulphate	100	150 Rs.		
	Coagulant Natural Moringa Okra Chemical Alum	CoagulantGramNatural coagulantMoringa100Okra100Chemical coagulantAlum100		







4. Conclusion

Various tests are carried out on the surface water to know its chemical physical a biological parameters like alkalinity, pH, hardness, TDS, TSS and MPN with different coagulants added in it.

With the help of results and cost we compare the natural and chemical coagulants; we conclude that the aluminum and moring powder is suitable for the treatment of the water.

Whereas the okra coagulant and ferric sulphate is totally not suitable for converting the surface water into the drinking water.

Percentage of effectiveness and cost:

- 1) Alum: 43%
- Moringa: 29% 2)
- 3) Okra: 15%
- 4) Ferric sulphate: 13%

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Fig. 6.